New Macrolactins from a Marine *Bacillus* sp. Sc026

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Bioassay-guided fractionation of the EtOAc extract of a marine *Bacillus* sp. Sc026 culture broth has led to the isolation of the known compound, macrolactin F (1), together with two new compounds, 7-O-succinyl macrolactin F (2) and 7-O-succinvl macrolactin A (3). The chemical structures of 1-3 were elucidated on the basis of spectral data analyses and literature data comparison. Compounds 1-3 exhibited antibacterial activity against Bacillus subtilis and Staphylococcus aureus.

As part of our interest in exploring the biologically active secondary metabolites of marine microorganisms obtained from the Thai marine environment, we have investigated several Bacillus strains isolated from sediments around Sichang Island (an island on the east coast in the Gulf of Thailand). We found antibacterial activity in the crude EtOAc extract of the marine Bacillus sp. Sc026 culture broth. Herein, we report the isolation, antibacterial activity, and structures of macrolactins from the marine Bacillus sp. Sc026, including the known compound, macrolactin F (1), together with two new compounds, 7-O-succinyl macrolactin F (2) and 7-O-succinyl macrolactin A (3). The macrolactins are polyene macrolides containing a 24membered lactone ring previously isolated from a taxonomically unidentified deep sea bacterium.^{1,2} So far, six macrolactins (A-F) have been chemically characterized, and only macrolactins A and E have been studied for total synthesis.^{3–6} Macrolactin A showed selective antibacterial activity, inhibited B16-F10 murine melanoma cancer cells in vitro assays, showed significant inhibition of mammalian Herpes simplex viruses (types I and II), and protected T-lymphoblast cells against human HIV viral replication.¹

The EtOAc extract from the 90-L culture broth of the marine *Bacillus* sp. Sc026 was subjected to repeated Si gel column chromatography to afford macrolactin F (1). Fractions possessing antibacterial activity were subsequently chromatographed by Sephadex LH-20, Si gel, and C₁₈ reversed-phase chromatography and preparative C₁₈ reversed-phase TLC, to yield two new macrolactins, 7-Osuccinyl macrolactin F (2) and 7-O-succinyl macrolactin A (3). The chemical structures of 1-3 were elucidated by extensive analyses of spectral data and comparison with known macrolactins.

The HRFABMS established the molecular formula of 1 as C24H34O5 by showing a molecular ion adduct peak at m/z 425.2326 [M + Na]⁺. The ¹H and ¹³C NMR data of 1 were identical with those of the known compound, macrolactin F, hence the identity of 1 was established.¹



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(3)

The molecular formula of C28H38O8 for 2 was determined by the HRFABMS showing a pseudomolecular ion peak at m/z 525.2490 [M + Na]⁺. Compound **2** has an extra C₄H₄O₃ unit compared with the molecular formula of 1. Extensive analyses of ¹H NMR, 2D ¹H-¹H COSY, and 2D TOCSY spectra of 2 indicated the similarity of the chemical structure to **1**. The downfield shift of H-7 (from δ 4.34 of **1**

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Table 1.	¹ H and	¹³ C NMR	Data of	Compounds	2 and 3	3 in	$CDCl_3$
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	2^a		3^b		
carbon	$\delta_{\rm C}$	$\delta_{ m H}$, mult. (J in Hz)	$\delta_{\rm C}$	$\delta_{\rm H}$, mult. (<i>J</i> in Hz)	
1	166.2		166.3		
2	117.9	5.53, d (11.6)	118.5	5.54, d (11.4)	
3	143.0	6.45, dd (10.9, 11.6)	141.8	6.48, dd (11.4, 11.6)	
4	130.0	7.19, dd (10.9, 15.5)	129.8	7.08, dd (11.6, 15.1)	
5	137.7	5.88, dt (7.5, 15.5)	137.7	5.59, dt (7.1, 15.1)	
6	38.4	2.48, 2H, m	39.4	2.55, 2H, dd (7.0, 7.1)	
7	72.5	5.41, m	72.3	5.54, m ^c	
8	130.8	5.62, dd (5.2, 15.2)	130.8	5.68, dd (3.4, 15.0)	
9	126.5	6.41, tdd (1.2, 10.9, 15.2)	124.9	6.59, dd (11.6, 15.0)	
10	130.5	6.04, dd (10.8, 10.9)	130.1	6.02 ^c	
11	127.8	5.44, dt (8.2, 10.8)	127.3	5.36, dt (5.5, 10.7)	
12	34.7	2.34, 2H, m ^c	34.5	2.40, m	
				2.92, dt (10.7, 13.0)	
13	68.0	4.01, m	70.6	4.09, m	
14	47.6	2.49, dd (7.9, 17.4)	39.1	1.58, m^c	
		2.55, dd (4.0, 17.4)		1.82, ddd (2.9, 9.5, 14.4)	
15	211.9		70.2	4.54, br t (7.1)	
16	43.6	2.40, 2H, t (7.3)	131.9	5.53^{c}	
17	27.0	2.20, 2H, m ^c	131.3	6.09 ^c	
18	128.5	5.32, m^c	130.0	5.99 ^c	
19	131.0	5.35, m ^c	135.2	5.59^{c}	
20	31.7	1.91, m	32.0	2.08, m	
		1.97, m		2.20, m	
21	24.9	1.34, 2H, m	24.5	1.45, 2H, m ^c	
22	35.0	1.44, m	34.9	1.55, 2H, m ^c	
		1.57, m			
23	70.9	4.95, m	71.5	4.98, m	
23-Me	20.0	1.18, 3H, d (6.4)	20.0	1.27, 3H, d (6.2)	
1′	171.0		170.8		
2'	28.9^{d}		28.8^{d}	2.65, 2H, m^d	
3′	29.6^{d}	} 2.58, 4H, br s	29.7^{d}	2.70, 2H, m^d	
4'	174.3		174.8		

^{*a*} Data were recorded at 500 MHz (¹H) and 75 MHz (¹³C). ^{*b*} Data were recorded at 300 MHz (¹H) and 75 MHz (¹³C). ^{*c*} Overlapping signals. ^{*d*} Assignments may be interchangeable.

to δ 5.41 of **2**) and the additional four equivalent protons (H_2-2', H_2-3') at δ 2.58 were the main spectral differences. The ¹³C NMR spectrum of **2** was also similar to that of **1**, with additional carbon signals at δ 28.9 (d), 29.6 (d), 171.0 (s), and 174.3 (s). The HMQC spectrum of 2 revealed connections of the four protons (H-2' and H-3') to the methylene carbons at δ 28.9 and 29.6. The HMBC spectrum of 2 demonstrated long-range correlations of the methylene protons (H-2' and H-3') to the carbonyl carbons at δ 171.0 (C-1') and 174.3 (C-4'). These observed spectral data were indicative of the presence of the succinic acid half-ester moiety in 2, which was supported by the IR spectrum of 2 exhibiting intensely broad hydroxyl absorption from 3500 to 2500 cm⁻¹, carboxylic carbonyl absorption at 1730 cm⁻¹, and ester carbonyl absorption at 1710 cm⁻¹. Placement of the succinic acid group at the C-7 hydroxyl group was confirmed by the downfield shift of H-7 (δ 5.41) in 2 compared to the shift of H-7 (δ 4.34) in ${\bf 1}$ and also by the long-range correlation of H-7 to the carbonyl carbon at δ 171.0 (C-1') in the HMBC spectrum of 2. Compound 2 was therefore 7-O-succinyl macrolactin F, a new derivative of the macrolactins.

As determined by the HRFABMS showing a pseudomolecular ion peak at 525.2421 [M + Na]⁺, the molecular formula of **3**, C₂₈H₃₈O₈, was the same as that of **2**. The appearance of the succinic acid half-ester fragment in **3** was defined by 2D NMR studies of ¹H⁻¹H COSY, HMQC, and HMBC spectra. The ¹H NMR resonances at δ 2.65 and 2.75 and the ¹³C NMR resonances at δ 28.8, 29.7, 170.8, and 174.8 were assigned to the succinyl group. The attachment of the succinyl group at the C-7 hydroxyl group was also secured by the downfield shift of H-7 at δ 5.54. Further 2D NMR analyses suggested that the macrolide ring of **3** was identical with that of macrolactin A.¹ The absence of the ketone carbon at δ 211.9 (C-15) and the presence of the oxymethine carbon at δ 70.2 (C-15), the olefinic carbons at δ 131.9 (C-16) and 131.3 (C-17), the oxymethine proton at δ 5.53 (H-15), and the olefinic protons at δ 6.09 (H-16) and 5.99 (H-17) revealed the characteristic allylic alcohol functionality at C-15 as in macrolactin A. Compound **3** was then assigned as a new macrolactin derivative, 7-*O*-succinyl macrolactin A.

Compounds **2** and **3** showed optical rotations similar to those of their parent macrolactins, F and A, respectively.² It can therefore, be assumed that both new macrolactins have the same stereochemistry as their respective parent molecules. Compounds **1**–**3** exhibited antibacterial activity against *B. subtilis* and *S. aureus* (inhibition zones of 8–28 mm at 50–100 μ g/disk).

Experimental Section

General Experimental Procedures. The 300-MHz NMR experiments were performed on a Bruker AVANCE DPX-300 FT-NMR spectrometer, while the 500-MHz NMR data were obtained on a JEOL JNM-A500 NMR spectrometer, and chemical shifts (ppm) were referenced relative to the residual undeuterated solvent signals. The number of attached protons for the ¹³C NMR signals was determined from DEPT experiments.7 Proton-detected heteronuclear correlations were measured using HMQC⁸ (optimized for ${}^{1}J_{HC} = 145$ Hz) and HMBC⁹ (optimized for ${}^{n}J_{HC} = 8$ Hz) pulse sequences. The HRFABMS spectra were obtained with JEOL JMS-HX110 mass spectrometer, operating at 10 kV ionization voltage and 3 kV acceleration voltage, using a mixture of glycerol and mnitrobenzyl alcohol as a matrix and NaCl as an alkali metal cation source. Optical rotations were measured on a Perkin-Elmer 341 polarimeter using a sodium lamp operating at 589 nm. IR and UV spectra were obtained on a Perkin-Elmer 2000

FT-IR spectrometer and a Spectronic 3000 UV spectrometer, respectively.

Microorganism. The bacterium (Sc026) was isolated from marine sediment collected nearby Sichang Island (at a 15-m depth), Chonburi, Thailand. The isolation medium was marine agar (MA). The characteristics of strain Sc026 were determined by the method described by Barrow and Feltham.¹⁰ The bacterium was Gram-positive, rod-shaped with subterminal spores, and motiled by peritrichous flagella. Colonies on a MA plate were opaque, circular, raised along the entire edge, and white. The bacterium showed the following characteristics: positive reaction to catalase and oxidase: reduction of nitrate: positive Voges-Proskauer reaction; hydrolysis of casein and starch, but exhibited negative reaction to urease, indole; utilization of citrate; and hydrolysis of hippurate. Growth occurred in media containing 0-8% NaCl (wt/vol). The bacterium could grow at temperatures ranging from 30 to 42 °C; however, the optimal temperature for growth was at 30 °C. Acid was produced from D-cellobiose, D-fructose, D-glucose, inositol, raffinose, salicin, sorbitol, sucrose, and D-trehalose, but not from L-arabinose, D-galactose, lactose, maltose, dmannitol, D-mannose, melibiose, melizitose, L-rhamnose, Dribose, sorbose, and D-xylose. On the basis of these available characteristics, this bacterium was identified as of the genus Bacillus.^{11,12} A specimen of Bacillus sp. Sc026 was deposited at the Department of Microbiology, Faculty of Pharmaceutical Sciences, Chulalongkorn University, Thailand.

Cultivation of Bacillus sp. Sc026. The bacterium was grown on MA plates and incubated at 30 °C for a day. A loopful of the bacterium was inoculated into a 500-mL Erlenmeyer flask containing 250 mL of marine broth and incubated on a rotatory shaker at 200 rpm, 30 °C, for 3 days.

Extraction and Isolation. The culture broth (90 L) of the marine Bacillus sp. Sc026 was extracted with EtOAc. The EtOAc layer was concentrated under reduced pressure to yield the crude extract (8 g), which was subjected to Si gel flash column chromatography using a gradient elution of CHCl₃-MeOH. Elution with 10% MeOH in CHCl₃ gave a 444-mg fraction, which was rechromatographed on a Si gel column using a gradient elution of CHCl3-EtOAc-MeOH to yield macrolactin F (1) (65 mg). The fraction (1.4 g) eluted with 30-40% MeOH in CHCl₃ was repeatedly subjected to gel filtration on a Sephadex LH-20 column using an isocratic elution of CHCl₃-MeOH-hexane (75:20:5) to give an antibacterial active fraction (616 mg). The fraction was subsequently chromatographed on a Si gel column using a gradient elution of CHCl3-MeOH, a C₁₈ reversed-phase column using a gradient elution of MeOH-H₂O, and finally by preparative C₁₈ reversed-phase TLC developed with a solvent system of MeOH-H₂O (4:1), yielding two new macrolactins, 7-O-succinyl macrolactin F (2) (7 mg) and 7-O-succinyl macrolactin A (3) (28 mg).

Antibacterial Assay. Antibacterial activity of the isolated compounds was performed using the agar diffusion assay

Macrolactin F (1): amorphous solid; $[\alpha]^{25}_{D} - 27.9^{\circ}$ (*c* 1.31, MeOH) (lit.¹ $[\alpha]_D$ – 30.1^{*o*}); ¹H and ¹³C NMR data in accordance with those of macrolactin F;1 HRFABMS m/z 425.2326 [M + Na^{+} (calcd for $C_{24}H_{34}O_5Na$, 425.2304).

7-O-Succinyl macrolactin F (2): amorphous solid; $[\alpha]^{25}$ _D -24.4° (*c* 0.50, MeOH); UV (MeOH) λ_{max} (log ϵ) 235 (4.17), 261 (4.23) nm; IR (CHCl₃) ν_{max} 3439, 1730, 1710, 1639, 1406 cm⁻¹; ¹H and ¹³C NMR data, Table 1; HRFABMS *m*/*z* 525.2490 [M + Na]⁺ (calcd for C₂₈H₃₈O₈Na, 525.2464).

7-O-Succinyl macrolactin A (3): amorphous solid; $[\alpha]^{25}$ _D -9.6° (c 0.18, MeOH); UV (MeOH) λ_{max} (log ϵ) 229 (4.57), 261 (4.18), nm; IR (CHCl₃) v_{max} 3295, 1730, 1700, 1667, 1557, 1410 cm⁻¹; ¹H and ¹³C NMR data, Table 1; HRFABMS *m*/*z* 525.2421 $[M + Na]^+$ (calcd for C₂₈H₃₈O₈Na, 525.2464).

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